

Amrital Raichand Mehta Degree College of Arts Diwali Maa Degree College of Science Mohanlal Raichand Mehta College of Commerce Padmashree (Dr.) R.T. Doshi Degree College of Computer Science Plot no.9, Sector -19, Airoli, Navi Mumbai

Program	B.Sc.
Year	2023-24
Eligibility Class	T.Y.B.Sc. (Chemistry)
Course	Bridge Course in Science
Number of Lectures	30
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#### Objectives of the Course

- To create strong problem-solving skills as a science student along with an understanding of the approach, methods of preparation and analysis in Chemistry for a successful career in advancing technology. Its science and design components prepare students to work as professional in Chemistry subject
- To build concrete foundation for their core branch as a thinker, inter disciplinary thoughts and socio- technical issues in field.
- To create advance applied knowledge of Chemistry as a subject of the foundation of Science.
- A breather, to prepare themselves before courses commence.

#### Course Outcome

- Student will be able to apply knowledge of basic theories of Chemistry in core subject of Science.
- Student will be able to create an experimental approach in applied theories of science in problem solving technical problems.
- Student will be able to associate natural phenomenon with technological advancement.
- It has helped students for enhancing critical thinking skills through self-exploration and class experiences.





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## SCIENCE DEPARTMENT

# BRIDGE COURSE SYLLABUS (2023-24)

#### LEVEL III

The Bridge Course for T. Y. B. Sc. (Chemistry) was divided into 5 modules. Following faculty members have undertaken the course:

• Dr. Harshada Niju: HN

• Asst. Prof. Sandhya Patil: SP

• Asst. Prof. Savita Kumari: SK

• Dr. Shweta Rathore: SR

Asst. Prof. Savita Kumari

Convener

Dr. Leena Sarkar

HOD (Chemistry) & Principal

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Dr. R.T. DOSHI COLLEGE OF COMPUTER SCIENCE
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#### SCIENCE DEPARTMENT

#### **BRIDGE COURSE SYLLABUS (2023-24)**

#### LEVEL III

	MODULE-1	6Hrs.
1.1.	Molecular Spectroscopy: Terms involved, Beer-Lamberts law, spectrum of	H.N.
	radiations. Types of spectra: rotational, vibrational, rotational-vibrational	
1.2.	Chemical Thermodynamics: Basic concepts and terms involved.	H.N.
	Chemical Kinetics: Terms, rate of reaction, rate constant, order of the reaction	
1.3.	Nuclear Chemistry: Types of radiation, radioactive elements, law of	H.N.
	disintegration	
1.4	Surfactant chemistry: absorption & adsorption differences, types of adsorption	H.N.
	Colloidal state: Particle size, true solution, suspensions, colloidal with examples	
1.5	Concept of concentration	H.N.
1.6	Surface Chemistry	H.N.
	MODULE-2	6Hrs.
2.1.	Symmetry and elements of symmetry 3L	S.R.
2.2	M.O.T of homoatomic molecules 3L	S.R.
	MODULE-3	6Hrs.
3.1.	Concepts of writing organic reaction mechanism, use of curved arrows	S.K
3.2.	Reaction Mechanism: Types of reactions, Electronic effects, Directing effects,	S.K
	types of reagents	
3.3	Basic terminologies in stereochemistry	S.K
3.4	IUPAC Nomenclature for organic compounds	S.K.
5.5	Concept of Green Chemistry	S.K.
.6	Introduction to spectroscopy	S.K

The same and the same of the same of	MODULE-4	6Hrs.
4.1	Sampling: Purpose, difficulties encountered	S.P
4.2	Methods of sampling: solids, liquids and gases	S.P
4.3	Solvent Extraction: Principle, batch and continuous extraction	S.P
4.4	Separation methods: chromatography	S.P
4.5	Chemical Calculation: molality, normality	S.P.
4.6	Chemical Calculation: molarity, Percent composition	S.P.
	MODULE-5	6Hrs.
5.1	General Introduction to Drug chemistry, Drugs from nature	S.R.
5.2	Basic terminologies used in drug chemistry	S.R.
5.3	Classification of drugs	S.R.
5.4	Introduction to dye Chemistry- Historical Development	S.K
5.5	Chemistry of colour in nature: Occurrence, chemical name & structure	S.K
5.6	Aromatic Electrophilic substitution reactions: Nitration, sulphonation,	S.K
	halogenations, alkylation, acylation, diazotization	

Asst. Prof. Savita Kumari

Convener

Dr. Leena Sarkar

HOD (Chemistry) & Principal

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# SCIENCE DEPARTMENT BRIDGE COURSE SYLLABUS (2023-24) List of Enrolled Students

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S. No.	NAME OF STUDENT
1	Omkar Nanavare
2	Madavi Kanse
3 .	Avneesh Yadav
4	Ritika Giri
5	Pratiksha Borde
6	Parul Singh
7	Sanika Pote
8	Pooja Shah
9	Riya Vishwakarma
10	Nikita Lagade
11	Jayawant Lotale

Asst. Prof. Savita Kumari

Convener

Dr. Leena Sarkar

HOD (Chemistry) & Principal

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#### SCIENCE DEPARTMENT BRIDGE COURSE SYLLABUS (2023-24) Assessment Record

S. No.	NAME OF STUDENT	TEST-1 14 Mks. (II.N)	TEST-2 20 Mks. (S.P)	TEST-3 20 Mks. (S.K)	TEST-4 10 Mks. (S.R)	Total out of 64
	Omkar Nanavare	A	14	16	3	33
2	Madavi Kanse	Λ	08	11	2	21
3	Avneesh Yadav	4	11	13	1	30
4	Ritika Giri	9	16	14	4	43
5	Pratiksha Borde	Α	10	11	1	22
6	Parul Singh	A	Α	A	A	A
7	Sanika Pote	Λ	Α	12	Λ	12
8	Pooja Shah	A	Α	Α	Α	A
9	Riya Vishwakarma	9	15	14	4	38
10	Nikita Lagade	7	12	14	1	34
11	Jayawant Lotale	A	Α	A	Α	A

Asst. Prof. Savita Kumari

Convener

HOD (Chemistry) & Principal

Dr. Leena Sarkar

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Bridge Course-Timetable 13<sup>th</sup> -19<sup>th</sup> June 2023 Class: T.Y.B.Sc.

Time	Day									
DOMESTIC MESTING THE CONTROL OF THE PROPERTY OF THE SECOND	TAGNIDAY	TUESDAY	WEDNESDAY	THURSDAY	FRIDAY	SATURDAY				
Day	MONDAY	TUESDAT			(H.N.)	(S.P.)				
9.00 to	(S.R.)	(S.P)	(S.P.)	(S.K)	(11.11.1)	(2.11)				
10.00 am	(6.17)	(S.P)	(S.R.)	(S.P.)	(S.R.)	(H.N.)				
10.00 to 11.00 am	(S.K)	(3.1)	(Birti)		(0.10)	(S.R.)				
11.20 to	(S.P.)	(S.K)	(H.N.)	(H.N.)	(S.K)	(5.K.)				
12.20 pm		(G.D.)	(5 17)	(S.R.)	(H.N.)	(S.K.)				
12.20 to	(H.N.)	(S.R.)	(S.K)	(31211)						
1.20 pm		Aggignment	Assignment	Assignment	Assignment	Assignment				
1.30 to	Assignment	Assignment	(S.R)	(S.K)	(( S.K)	(H.N)				
2.30 p.m.	(S.P)	(S.R)	(5.10)	, ,						

H.N – Dr. Harshada Niju S.P – Mrs. Sandhya Patil S.K – Miss Savita Kumari S.R – Dr. Shweta Rathore

Sign of Class teacher -

Sign of H.O.D-

Sign of Principal-

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#### SCIENCE DEPARTMENT BRIDGE COURSE SYLLABUS (2023-24) Summary Report

The goals of the Bridge course were accomplished successfully. Before the course begins, the course was designed to assist students understand the fundamental concepts of core themes. The course aided students in gaining a better understanding of what to expect at their new academic level and in settling into their new program. It aided pupils in gaining self-assurance and learning new talents that they may utilize in their future academic pursuits. Overall, it gave the learner the opportunity to:

- Develop strong problem-solving skills as a science student
- Develop a solid foundation for their core branch as a thinker, including thoughts from other disciplines and socio-technical challenges in the field
- To develop advanced practical knowledge of several scientific subjects

Asst. Prof. Savita Kumari

Convener

HOD (Chemistry) & Principal

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JNAN VIKAS MANDAL'S DEGREE COLLEGE
NAAC RE-ACCREDITED 'A' GRADE (CGPA 3:33)
Plot No. 9,5ector-19,Alroll Navi Mumbal

# Certificate

This is to certify that «Name» of Class T.Y.B.Sc. (Chemistry) has participated in 30 hours bridge course organized by Department of Science, JVM's Degree College.

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Ms. Savita Kumari Convener

Leena

Dr. Leena Sarkar Principal

Asst. Prof. Savita Kumari

Convener

Dr. Léena Sarkar

HOD (Chemistry) & Principal

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1	Riya vishwakam	CHEM	Ruper	Riyav	Rigar	Riya
2	Ritika Givi	CHEM	Ritika	Pilite	Petito	_P. tik
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4	Nikita Lagade	cham	Wirly.	Wkilg.	Dikity.	Dikila
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	JVM'S MEHTA	DEGREE	COLLEG	E			1
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r. No.	Name of the Student	Subject	Lec -1	Lec- 2	Lec-3	Lec-4	1
1	Nikita Lagade.	chem	Mille	WKity		- 1: -	10
2	Riya visheralom	chem	Ryd	Rugy		Duje	4
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3	Nikita Lagade	Thom	mikely	MKH	014	CAR
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5	Ritika giri	Chem		Ritils	Glike	PHI
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	JVM'S MEHTA	DEGREE	COLLEG	E		
	Bridge Co	urse :- 20	23-24			
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		Date: 19	06	23		
r. No. N	lame of the Student	Subject	Lec -1	Lec- 2	Lec-3	Lec-4
	Riya vishevakam	Clum	Ryan	Rigar	Ruga	Ruy
2	Nikito Lagade.	Chem	Akita	Mikis	dove	FRIK
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1	Madavi Kanase	chem	moore	Unon	Mach	Molay
2	Ritika Cruini	chem	Pitika	Pitika	Patik	Pilito
3	Mikita Lagade.	chem	Mikity	wikita-	Alkha	PENT
4	Avancest radar	Chom	70	A	A	
5	Pratiksha Borde		Bosda	Forde	Roods	30.19 B
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